

Fig. 2. The volume change on melting of He³ and He⁴ at low pressures. The large circles represent measurements made with the small, heavy-walled cell, while the small circles represent those made with the large, thin-walled cell.

 dP_m/dT_m . Similar properties for $\mathrm{He^3}$ are given in Table II. Maximum errors are estimated to be 0.5 percent for ΔV_m , 0.1 percent for V_f , and 1 percent for ΔS_m . In Table III are presented the melting parameters for the solid-solid transition of $\mathrm{He^3}$. Results of the high-pressure, room-temperature gas-density determinations for $\mathrm{He^4}$ and $\mathrm{He^3}$ are given in Table IV.

As in the case of N_2 (15), the ΔV_m data were fitted to the equation

$$\Delta V_m = A - B \log_{10}(P_m + C) \tag{1}$$

by the method of least squares. For He 3 two sets of constants were needed—one for the region below the triple point and the other for the region above. It was not possible to fit the He 4 ΔV_m data to Eq. (1) over the full pressure range studied. However, for the purpose of interpolation, a fit was made from 175 to 3555 kg/cm 2 . The constants in Eq. (1) for the various solids are presented in Table V. Listed also are the pressure range and rms deviation in ΔV_m .

The melting curve data at low pressure were fitted by the method of least squares to analytical expressions of the form,

$$P = A' + B'T + C'T^{2} + D'T^{3} + E'T^{4}.$$
 (2)